

**ADDITIVES TO PREVENT DEGRADATION OF ALKYL-HYDROGEN**  
**SILOXANES**

5   **RELATED APPLICATION**

This application claims priority from Provisional Patent Application Serial No. 60/411,651 filed on September 18, 2002, pending.

10   **BACKGROUND OF THE INVENTION**

1. Field of the Invention

15       The present invention relates generally to siloxane compositions that exhibit stability to moisture and oxygen. More particularly, the present invention is directed to organohydrosiloxanes stabilized with one or more antioxidant compounds.

2. Description of the Prior Art

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1,3,5,7-Tetramethylcyclotetrasiloxane (TMCTS) is one example of an organohydrosiloxane (i.e., constructed of [-HSiR-O-] units where R is alkyl, and in this case, methyl) that can be used for specialty polymer formation. TMCTS is an example of a cyclic organohydrosiloxane, and it  
25   has the ability to undergo ring opening polymerization. It also has reactive Si-H bonds that can be used for adding functionality to the resulting polymer. Due to these unique properties, the semiconductor industry recently chose TMCTS as a new precursor for the plasma enhanced chemical vapor deposition (PECVD) of Low K dielectric films on  
30   semiconductor wafers by chemical vapor deposition.

The introduction of TMCTS and similar cyclic organohydrosiloxanes into the semiconductor fab has been causing serious manufacturing

challenges because these materials can polymerize unpredictably. This behavior can be very costly and time consuming if polymerization occurs inside liquid or vapor chemical delivery lines where the only remedy is to disassemble and replace the affected lines. The semiconductor industry  
5 demands a stable, predictable and reliable product, and this behavior is unacceptable for high volume semiconductor manufacturing. Therefore, it is necessary to find a means to stabilize TMCTS to ensure that the product does not polymerize during transport from the chemical supplier to the end-use process, even after exposure to various conditions.

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TMCTS and other alkyl-hydrogen linear or cyclic siloxanes having Si-H bonds can react with oxygen or other components of air. This reaction can cause significant degradation of the siloxane, and when cyclic materials are involved, complete polymerization can eventually result, even  
15 after brief air exposure at room temperature.

Degradation of the chemical reduces shelf life, lowers product quality, and can even cause greater problems such as equipment downtime if the material polymerizes in chemical processing equipment.  
20 Safety issues are also a concern if pressurized chemical lines and valves become blocked.

For the semiconductor industry, degradation of an organohydrosiloxane product to higher molecular weight species causes  
25 additional problems with chemical use. The most common method to introduce high boiling point (e.g., > 50 °C) materials into a processing tool is by directly injecting a vapor or mist of chemical into a heated "carrier" gas stream that flows into a process chamber below ambient pressure. The heated stream of carrier gas vaporizes the product, and the product vapor  
30 is transported to the process chamber where the siloxane is transformed into a carbon doped silicon oxide film. However, this technique usually involves a single, pure species in order to have homogeneous product vapor inside the equipment. If high molecular weight species are present,

as found in degraded organohydrosiloxanes, they do not readily vaporize in the heated gas stream, but instead simply collect on the internal surfaces of the process equipment. This leads to several problems, including, for example, clogged transfer lines and valves and particle formation on a  
5 wafer.

There have been prior art attempts to stabilize siloxane and silane-containing compositions. U.S. Patent No. 5,177,142 to Stein et al. discloses silicone release coating compositions comprising a  
10 polydiorganosiloxane containing unsaturated group, a crosslinking agent containing Si-H groups, and a hydrosilation catalyst. The catalyst is inhibited by incorporating at least one free radical inhibitor, such as alkoxyphenols, in the composition. As a result, the compositions have a reduced tendency to bond to acrylic adhesives.

15 U.S. Patent No. 5,556,901 to Bryson discloses a polysilane-based composition used particularly for producing silicon carbide. The composition comprises polysilane chains capable of being degraded by oxidation and the formation of polysiloxane-type products, and at least one  
20 antioxidant system. The antioxidant system greatly reduces the sensitivity to air of the polysilanes.

What is lacking in the prior art is a composition, like the present invention, comprising organohydrosiloxanes that exhibits improved stability  
25 to moisture and oxygen, for use in the semiconductor field. The stable composition of the present invention significantly extends the shelf life of organohydrosiloxane products and provides greater flexibility in handling these products in chemical processes or semiconductor manufacturing. The resulting stabilization of siloxanes prevents the possibility of complete  
30 polymerization (i.e., solidification) of product in chemical delivery lines or valves, which lowers equipment maintenance and costs and reduces time the machinery is out of production. In addition, reducing the formation of higher molecular weight compounds allows homogeneous vaporization of

the product without concern for the gradual deposition of higher molecular weight compounds in vapor delivery lines leading to more consistent, higher quality deposited films.

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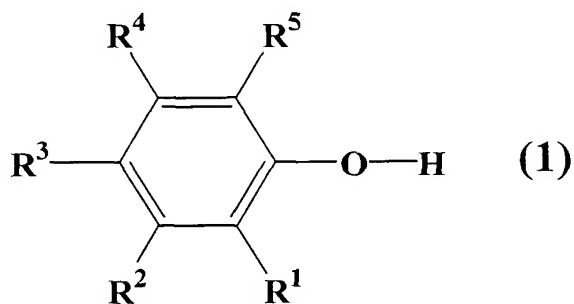
## SUMMARY OF THE INVENTION

The invention provides an organohydrosiloxane composition comprising:

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- a. one or more organohydrosiloxane compounds, each having at least one [-HSiR-O-] unit, wherein R = C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl; and
- b. an antioxidant compound of Formula (1),

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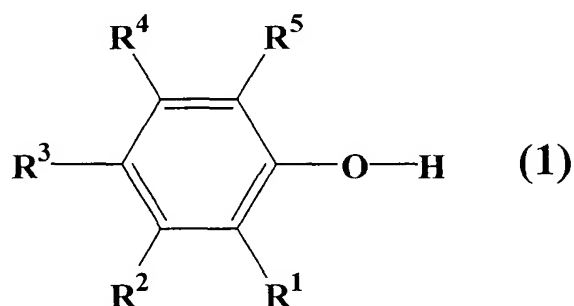


wherein the antioxidant compound is a phenolic compound and is present in an amount between about 1 ppm to about 5000 ppm, and wherein R<sup>1</sup> through R<sup>5</sup> are each independently H, OH, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

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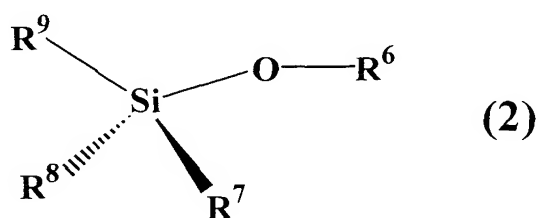
The invention also provides an organohydrosiloxane composition comprising:

- a. one or more organohydrosiloxane compounds, each having at least one [-HSiR-O-] unit, wherein R = C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl;
- 5 b. an antioxidant compound of Formula (1),



wherein the antioxidant compound is a phenolic compound and is present in an amount between about 1 ppm to about 5000 ppm, and wherein R<sup>1</sup> through R<sup>5</sup> are each independently H, OH, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl; and

- 15 c. an alkoxysilane of Formula (2),



20 wherein the alkoxysilane is present in an amount between about 1 ppm and about 5000 ppm; and wherein R<sup>6</sup> is a C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl or substituted or unsubstituted aryl; and R<sup>7</sup>, R<sup>8</sup>, and R<sup>9</sup> are independently H, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

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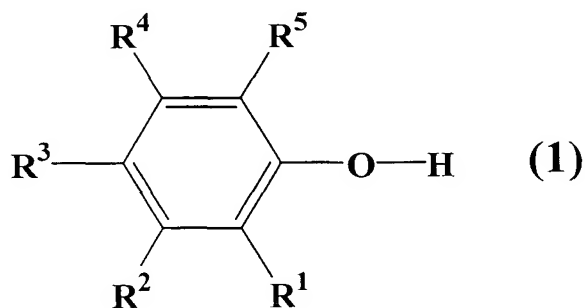
In addition to the above compositions, the present invention also provides a process for forming an oxide layer on a substrate comprising the steps of:

- 5           a.     introducing an organohydrosiloxane composition into a gas stream, thereby forming a process vapor;
- b.     contacting a surface of the substrate with the process vapor; and
- 10          c.     decomposing the process vapor, thereby forming the oxide layer on the substrate,

wherein the organohydrosiloxane composition comprises:

- 15           a.     one or more organohydrosiloxane compounds, each having at least one  $[-HSiR-O-]$  unit, wherein  $R = C_1-C_{18}$  linear, branched, or cyclic alkyl,  $C_1-C_{18}$  linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl; and
- b.     an antioxidant compound of Formula (1),

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- 25           wherein the antioxidant compound is a phenolic compound and is present in an amount between about 1 ppm to about 5000 ppm, and wherein  $R^1$  through  $R^5$  are each independently H, OH,  $C_1-C_{18}$  linear, branched, or cyclic alkyl,  $C_1-C_{18}$  linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

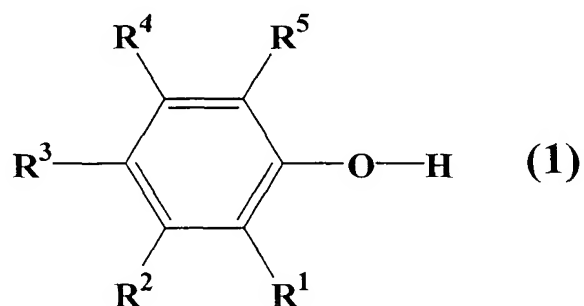
Another process for forming an oxide layer on a substrate according to the present invention comprises the steps of:

- 5           a.     introducing an organohydrosiloxane composition into a gas stream, thereby forming a process vapor;
- b.     contacting a surface of the substrate with the process vapor; and
- 10          c.     decomposing the process vapor, thereby forming the oxide layer on the substrate,

wherein the organohydrosiloxane composition comprises:

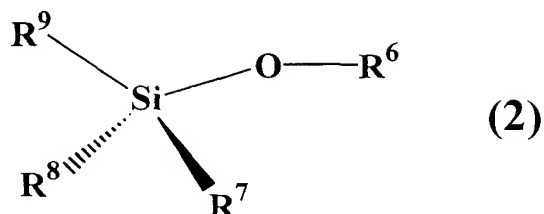
- 15           a.     one or more organohydrosiloxane compounds, each having at least one [-HSiR-O-] unit, wherein R = C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl;
- b.     an antioxidant compound of Formula (1),

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- 25           wherein the antioxidant compound is a phenolic compound and is present in an amount between about 1 ppm to about 5000 ppm, and wherein R<sup>1</sup> through R<sup>5</sup> are each independently H, OH, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl; and

c. an alkoxy silane of Formula (2),



wherein the alkoxy silane is present in an amount between about 1 ppm to about 5000 ppm; and wherein R<sup>6</sup> is a C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl or substituted or unsubstituted aryl; and R<sup>7</sup>, R<sup>8</sup>, and R<sup>9</sup> are independently H, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

## BRIEF DESCRIPTION OF DRAWING

Figure 1 is a representation of a film deposition tool used in the semiconductor industry for use with the compositions according to the present invention.

## DETAILED DESCRIPTION OF THE INVENTION

The compositions of present invention exhibit stability and significantly extend the shelf life of organohydrosiloxane products, allowing for greater flexibility in the handling of these products in chemical processes or semiconductor manufacturing. The resulting stabilization of siloxanes prevents the possibility of complete polymerization (i.e., solidification) of product in chemical delivery lines or valves, which lowers equipment maintenance and costs and reduces the time which the machinery is out of production. In addition, reducing the formation of

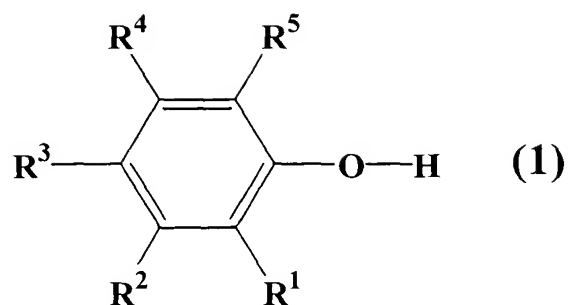


higher molecular weight compounds allows homogeneous vaporization of the product without concern for the gradual deposition of higher molecular weight compounds in vapor delivery lines leading to more consistent, higher quality deposited films. It is furthermore an objective of this invention to provide a process of use of stabilized organohydrosiloxane compositions.

One important aspect of the present invention is that films deposited with the compositions of the present invention have been found to have properties similar to and/or better than those films deposited using existing compositions. Therefore, the compositions of the present invention can seamlessly replace existing compositions without requiring process changes and/or experiencing undesirable performance changes.

In one embodiment of the present invention, the organohydrosiloxane composition comprises:

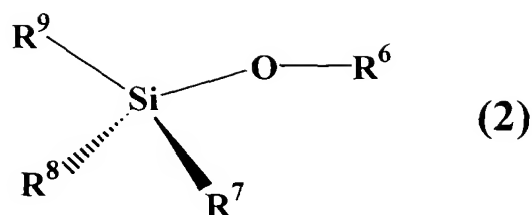
- A) one or more organohydrosiloxane compounds, each having at least one [-HSiR-O-] unit, wherein R = C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl; and
- B) an antioxidant compound of Formula (1), wherein the antioxidant compound is a phenolic compound and is present in a concentration from about 1 ppm to about 5,000 ppm and wherein R<sup>1</sup> through R<sup>5</sup> can each independently be H, OH, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.



In another embodiment of the present invention, the  
 5 organohydrosiloxane composition comprises:

- a) one or more organohydrosiloxane compounds, each having at least one [-HSiR-O-] unit, wherein R = C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy, or substituted or  
 10 unsubstituted aryl;
- b) an antioxidant compound of Formula (1), wherein the antioxidant compound is a phenolic compound and is present in a concentration from about 1 ppm to about 5,000 ppm and wherein R<sup>1</sup> through R<sup>5</sup> can each independently be H, OH, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic  
 15 alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl; and
- c) an alkoxysilane of formula (R<sup>6</sup>O)SiR<sup>7</sup>R<sup>8</sup>R<sup>9</sup> shown in Formula (2), wherein the alkoxysilane compound is present in a concentration from about 1 ppm to about 5,000 ppm, and R<sup>6</sup> is a C<sub>1</sub>-C<sub>18</sub> linear,  
 20 branched, or cyclic alkyl or substituted or unsubstituted aryl and R<sup>7</sup>, R<sup>8</sup>, and R<sup>9</sup> are independently H, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

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The present invention is also directed to a process of using an  
 5 organohydrosiloxane composition for forming a layer of carbon-doped  
 silicon oxide on a wafer.

In an embodiment of the present invention, the process comprises  
 the steps of :

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- a) providing an organohydrosiloxane composition in a container, a film  
 deposition tool, a means for connecting the container and the film  
 deposition tool, a stream of gas and a substrate within the film  
 deposition tool;
- 15 b) introducing the organohydrosiloxane composition into a gas stream  
 thereby forming a process vapor;
- c) transporting the vapor of the organohydrosiloxane composition to  
 the substrate within the film deposition tool; and
- d) using one or more means, such as plasma, heat or a chemical  
 20 reaction, to decompose the organohydrosiloxane composition in  
 order to form a layer of carbon-doped silicon oxide on the substrate,

wherein the organohydrosiloxane composition comprises:

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- 1) one or more organohydrosiloxane compounds, each having  
 at least one [-HSiR-O-] unit, wherein R = C<sub>1</sub>-C<sub>18</sub> linear,  
 branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic  
 alkoxy, or substituted or unsubstituted aryl; and
- 2) an antioxidant compound of Formula (1), wherein the  
 30 antioxidant compound is a phenolic compound and is present

in a concentration from about 1 ppm to about 5,000 ppm and wherein  $R^1$  through  $R^5$  can each independently be H, OH,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkyl,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

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In another embodiment of the present invention, the process of using an organohydrosiloxane composition for forming a layer of carbon-doped silicon oxide on a wafer comprises the steps of:

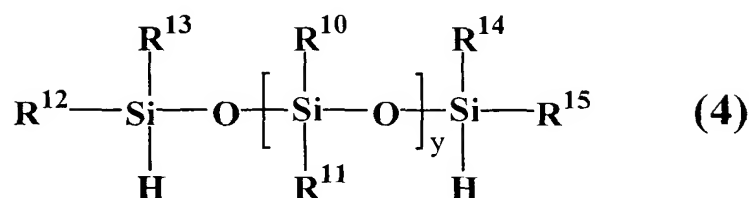
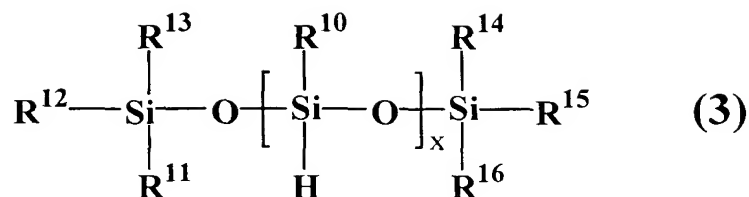
- 10 a) providing an organohydrosiloxane composition in a container, a film deposition tool, a means for connecting the container and the film deposition tool, a stream of gas and a substrate within the film deposition tool;
- b) introducing the organohydrosiloxane composition into a gas stream
- 15 thereby forming a process vapor;
- c) transporting the vapor of the organohydrosiloxane composition to the substrate within the film deposition tool; and
- d) using one or more means, such as plasma, heat or a chemical reaction, to decompose the organohydrosiloxane composition in
- 20 order to form a layer of carbon-doped silicon oxide on the substrate,

wherein the organohydrosiloxane composition comprises:

- 1) one or more organohydrosiloxane compounds, each having at least one  $[-HSiR-O-]$  unit, wherein  $R = C_1$ - $C_{18}$  linear, branched, or cyclic alkyl,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl;
- 25 2) an antioxidant compound of Formula (1), wherein the antioxidant compound is a phenolic compound and is present in a concentration from about 1 ppm to about 5,000 ppm and wherein  $R^1$  through  $R^5$  can each independently be H, OH,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkyl,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl; and
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- 3) an alkoxysilane of formula  $(R^6O)SiR^7R^8R^9$  shown in Formula (2), wherein the alkoxysilane compound is present in a concentration from about 1 ppm to about 5,000 ppm, and  $R^6$  is a  $C_1$ - $C_{18}$  linear, branched, or cyclic alkyl or substituted or unsubstituted aryl and  $R^7$ ,  $R^8$ , and  $R^9$  are independently H,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkyl,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl.

The organohydrosiloxane may be linear or cyclic. Suitable examples of linear organohydrosiloxane compounds of the present invention are those described by Formula (3) or alternatively in Formula (4). In Formulas (3) or (4),  $R^{10}$  can be  $C_1$ - $C_{18}$  linear, branched, or cyclic alkyl,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl, and  $R^{11}$  through  $R^{16}$  can each independently be H,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkyl,  $C_1$ - $C_{18}$  linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl, x can range from 1 to about 20 and y can range from 0 to about 20. Additionally, x can equal 0 when at least one of  $R^{11}$  through  $R^{16}$  equals H. Examples of Formulas (3) and (4) include, but are not limited to, having  $R^{10}$  through  $R^{16}$  equal methyl, ethyl, propyl, butyl, phenyl, methylphenyl, cyclohexyl, isopropyl and additionally H for  $R^{11}$  through  $R^{16}$ , where y can range from about 0 to about 10 and x can range from about 1 to about 10 or instead equal 0 if one of  $R^{11}$  through  $R^{16}$  equals H. Preferred examples of Formulas (3) and (4) include, but are not limited to, having  $R^{10}$  through  $R^{16}$  equal methyl, ethyl, propyl, butyl, cyclohexyl and additionally H for  $R^{11}$  through  $R^{16}$ , where x can range from about 1 to about 8 and y can range from 0 to about 8. More preferred examples of Formulas (3) and (4) include, but are not limited to, having  $R^{10}$  through  $R^{16}$  methyl, ethyl, cyclohexyl and additionally H for  $R^{11}$  through  $R^{16}$ , where x can range from about 1 to about 4 and y can range from about 1 to about 5. For Formulas (3) and (4) used in semiconductor applications, methyl is most preferred for  $R^{10}$ , methyl or H is most preferred for  $R^{11}$  through  $R^{16}$ , and x can range from about 3 to about 4 while y can range from about 3 to about 4.



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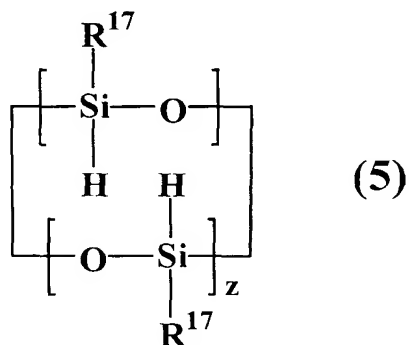
Suitable examples of linear organohydrosiloxanes described by Formula (3) include 1,1,1,3,3-pentamethyldisiloxane, 1,1,1,3,3-pentaethyldisiloxane, 1,1,1,3,3-pentaphenyldisiloxane, 1,1,1,3,3-penta(4-methylphenyl)disiloxane, 1,1,5,5-tetramethyl-3-ethyltrisiloxane, 1,1,5,5-tetraethyl-3-methyltrisiloxane, 1,1,3,5,5-pentamethyltrisiloxane, 1,1,3,5,5-pentaethyltrisiloxane, 1,1,3,5,5-pentaphenyltrisiloxane, 1,1,3,5,5-penta(4-methylphenyl)trisiloxane, 1,1,1,5,5,5-heptamethyl-3-ethyltrisiloxane, 1,1,1,5,5,5-heptaethyl-3-methyltrisiloxane, 1,1,1,3,5,5,5-heptamethyltrisiloxane, 1,1,1,3,5,5,5-heptaethyltrisiloxane, 1,1,1,3,5,5,5-heptaphenyltrisiloxane, 1,1,1,3,5,5,5-hepta(4-methylphenyl)trisiloxane, 1,1,3,5,7,7-hexamethyltetrasiloxane, 1,1,3,5,7,7-hexaethyltetrasiloxane, 1,1,3,5,7,7-hexaphenyltetrasiloxane, 1,1,3,5,7,7-hexa(4-methylphenyl)tetrasiloxane, 1,1,1,3,5,7,7,7-octamethyltetrasiloxane, 1,1,1,3,5,7,7,7-octaethyltetrasiloxane, 1,1,1,3,5,7,7,7-octaphenyltetrasiloxane, 1,1,1,3,5,7,7,7-octa(4-methylphenyl)tetrasiloxane, 1,1,3,5,7,9,9-heptamethylpentasiloxane, 1,1,3,5,7,9,9-heptaethylpentasiloxane, 1,1,3,5,7,9,9-heptaphenylpentasiloxane, 1,1,3,5,7,9,9-hepta(4-methylphenyl)pentasiloxane, 1,1,1,3,5,7,9,9,9-nonamethylpentasiloxane, 1,1,1,3,5,7,9,9,9-nonaethylpentasiloxane, 1,1,1,3,5,7,9,9,9-nonaphenylpentasiloxane, 1,1,1,3,5,7,9,9,9-nona(4-

- methylphenyl)pentasiloxane, 1,1,3,5,7,9,11,11-octaamethylhexasiloxane, 1,1,3,5,7,9,11,11-octaethylhexasiloxane, 1,1,3,5,7,9,11,11-octaphenylhexasiloxane, 1,1,3,5,7,9,11,11-octa(4-methylphenyl)hexasiloxane, 1,1,1,3,5,7,9,11,11,11-decamethylhexasiloxane, 1,1,1,3,5,7,9,11,11,11-decaethylhexasiloxane, 1,1,1,3,5,7,9,11,11,11-decaphenylhexasiloxane, 1,1,1,3,5,7,9,11,11,11-deca(4-methylphenyl)hexasiloxane.

- Suitable examples of linear organohydrosiloxanes described by Formula (4) include 1,1-diethyl-3,3-dimethyldisiloxane, 1,1,3,3-tetramethyldisiloxane, 1,1,3,3-tetraethyltrisiloxane, 1,1,3,3-tetraphenyldisiloxane, 1,1,3,3-tetra(4-methylphenyl)disiloxane, 1,1,5,5-tetramethyl-3,3-diethyltrisiloxane, 1,1,5,5-tetraethyl-3,3-dimethyltrisiloxane, 1,1,3,3,5,5-hexamethyltrisiloxane, 1,1,3,3,5,5-hexaethyltrisiloxane, 1,1,3,3,5,5-hexaphenyltrisiloxane, 1,1,3,3,5,5-hexa(4-methylphenyl)trisiloxane, 1,1,3,3,5,5,7,7-octamethyltetrasiloxane, 1,1,3,5,7,7-octaethyltetrasiloxane, 1,1,3,3,5,5,7,7-octaphenyltetrasiloxane, 1,1,3,3,5,5,7,7,9,9-decamethylpentasiloxane, 1,1,3,3,5,5,7,7,9,9-decaethylpentasiloxane, 1,1,3,3,5,5,7,7,9,9-decaphenylpentasiloxane, 1,1,3,3,5,5,7,7,9,9-deca(4-methylphenyl)pentasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11-dodecaamethylhexasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11-dodecaethylhexasiloxane, 1,1,3,3,5,5,7,7,9,9,11,11-dodecaphenylhexasiloxane, and 1,1,3,3,5,5,7,7,9,9,11,11-dodeca(4-methylphenyl)hexasiloxane.

- Suitable examples of cyclic organohydrosiloxane compounds of the invention are those described by Formula (5). In Formula (5), each R<sup>17</sup> can independently be C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy, or substituted or unsubstituted aryl and z can range from about 2 to about 21. Examples of Formula (5) include, but are not limited to having R<sup>17</sup> equal to methyl, ethyl, propyl, butyl, phenyl, methylphenyl, cyclohexyl, or isopropyl, where z can range from about 2 to

about 21. Preferred examples of Formula (5) include having  $R^{17}$  equal to methyl, ethyl, propyl, butyl, or cyclohexyl, where  $z$  can range from about 2 to about 11. More preferred examples of Formula (5) include having  $R^{17}$  equal methyl, ethyl, or cyclohexyl, where  $z$  can range from about 2 to about 7. Most preferred examples of Formula (5) include having  $R^{17}$  equal methyl and  $z$  can range from about 2 to about 5 for use in semiconductor applications.



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Examples of cyclic siloxanes described by Formula (5) include 1,3,5-trimethylcyclotrisiloxane, 1,3,5-triethylcyclotrisiloxane, 1,3,5-triphenylcyclotrisiloxane, 1,3,5-tri(4-methylphenyl)cyclotrisiloxane, 1,3,5,7-tetramethylcyclotetrasiloxane, 1,3,5,7-tetraethylcyclotetrasiloxane, 1,3,5,7-tetraphenylcyclotetrasiloxane, 1,3,5,7-tetra(4-methylphenyl)cyclotetrasiloxane, 1,5-dimethyl-3,7-diethylcyclotetrasiloxane, 1,3-dimethyl-5,7-diethylcyclotetrasiloxane, 1,3,5,7,9-pentamethylcyclopentasiloxane, 1,3,5,7,9-pentaethylcyclopentasiloxane, 1,3,5,7,9-pentaphenylcyclopentasiloxane, 1,3,5,7,9-penta(4-methylphenyl)cyclopentasiloxane, 1,3,5,7,9,11-hexamethylcyclohexasiloxane, 1,3,5,7,9,11-hexaethylcyclohexasiloxane, 1,3,5,7,9,11-hexaphenylcyclohexasiloxane, 1,3,5,7,9,11-hexa(4-methylphenyl)cyclohexasiloxane, 1,5,9-trimethyl-3,7,11-triethylcyclohexasiloxane, and 1,3,5-trimethyl-7,9,11-triethylcyclohexasiloxane.

20

25



Cyclic organohydrosiloxanes, as in Formula (5), are preferred for use in the semiconductor industry due to their greater ability to form polymeric species when compared to linear compounds, as in Formulas (3) or (4). Preferred examples of cyclic organohydrosiloxanes include 1,3,5-trimethylcyclotrisiloxane, 1,3,5-triethylcyclotrisiloxane, 1,3,5,7-tetramethylcyclotetrasiloxane, 1,3,5,7-tetraethylcyclotetrasiloxane, 1,3,5,7,9-pentamethylcyclopentasiloxane and 1,3,5,7,9-pentaethylcyclopentasiloxane. 1,3,5-trimethylcyclotrisiloxane and 1,3,5,7-tetramethylcyclotetrasiloxane are most preferred for use in the semiconductor industry.

The properties of a layer deposited on a substrate, such as dielectric constant, film hardness and refractive index, are influenced by changing the composition of the chemistry that is fed into the film deposition tool. For example, the dielectric constant may range from about 2.0 to about 4.0, depending on the processing conditions used. Materials with a dielectric constant below about 2.7 are preferred for newer semiconductor devices. The layer properties can be tuned by changing the identity of the organohydrosiloxane, by using a different flow gas, or by using one or more different reactive gases. Another means to affect the layer properties is to use a combination of organohydrosiloxanes, or to combine an organohydrosiloxane with other silicon-containing molecules such as silanes, polysilanes, siloxanes, polysiloxanes and the like. The decision to combine organohydrosiloxanes or to use them in combination with other silicon compounds, and the proportion and method in which they are combined, is wholly dependent on the desired effect one wants to achieve. In such a combination, the relative ratios may range from about 1% to about 99% in the case of a two-component system and from about 1% to about 99% in a three-component system with the proviso that the sum of the three components equals about 100% of the composition excluding any additives.

Suitable phenolic antioxidants of the invention are described by Formula (1) where R<sup>1</sup> through R<sup>5</sup> can each independently be H, OH, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkyl, C<sub>1</sub>-C<sub>18</sub> linear, branched, or cyclic alkoxy or substituted or unsubstituted aryl. Examples of suitable R<sup>1</sup> through R<sup>5</sup> include H, OH, methyl, ethyl, propyl, iso-propyl, butyl, iso-butyl, tert-butyl, methoxy, ethoxy, propoxy, iso-propoxy, butoxy or tert-butoxy. Preferred examples of R<sup>1</sup> through R<sup>5</sup> in Formula (1) include H, OH, methyl, ethyl, methoxy, ethoxy, and tert-butyl. Most preferred examples are H and methoxy, where R<sup>1</sup>, R<sup>2</sup>, R<sup>4</sup> and R<sup>5</sup> are H and R<sup>3</sup> is methoxy.

10

Suitable examples of Formula (1) include, but are not limited to, phenol, hydroquinone, 4-methylphenol, 3-methylphenol, 2-methylphenol, 4-ethylphenol, 4-propylphenol, 4-iso-propylphenol, 4-butylphenol, 4-sec-butylphenol, 4-iso-butylphenol, 4-tert-butylphenol, 4-methoxyphenol, 3-methoxyphenol, 2-methoxyphenol, 4-ethoxyphenol, 4-propoxyphenol, 4-butoxyphenol, 2,4-di-tert-butylphenol, 2-(1-methylbutyl)phenol, 2-(benzyloxy)phenol, 2-tert-butyl-6-methylphenol, 3,4,5-trimethoxyphenol, 3-ethoxy-4-methylphenol, 4-benzyloxyphenol, 4-benzyl-2,6-di-tert-butylphenol, 2-(2-butenyl)phenol, 2-(4-methylbenzyl)phenol, 2,6-di-tert-butyl-4-methylphenol (BHT), 1,2-dihydroxybenzene, 2,4,6-tris-benzyloxyphenol, 2,4-dicyclohexyl-5-methylphenol and 6-tert-butyl-1,2-dihydroxybenzene. Preferred antioxidants of Formula (1) include phenol, 4-methylphenol, 3-methylphenol, 2-methylphenol, 4-methoxyphenol, 3-methoxyphenol, and 2-methoxyphenol. The most preferred antioxidant of Formula (1) is 4-methoxyphenol.

25

A suitable concentration of phenolic additive may be obtained from a range of concentrations, such as from about 1 ppm to about 5000 ppm. Suitable concentration ranges can be defined by defining a low end of suitable concentrations and a high end of suitable concentrations.

30

The low end of the suitable concentration range for the phenolic additive may be from about 1 ppm to about 200 ppm. Suitable

concentrations for the low end concentration range include, but are not limited to, 1 ppm, 5 ppm, 10 ppm, 25 ppm, 50 ppm, 100 ppm, 150 ppm, and 200 ppm.

5           The high end of the suitable concentration range is not critical for stabilization of the compositions, but may be limited by some considerations of the deposited film purity and solubility of the phenolic additive in the organohydrosiloxane. A suitable concentration range for the high end phenolic additive concentration may be from about 1000 ppm to  
10   about 5000 ppm. Examples of suitable concentrations for the high end include but are not limited to 1000 ppm, 1500 ppm, 2000 ppm, 2500 ppm, 3000 ppm, 3500 ppm, 4000 ppm, 4500 ppm, and 5000 ppm.

          An example of a suitable concentration of phenolic additive is a  
15   range from 1 to about 1000 ppm. A preferred concentration of phenolic additive is a range from about 10 to about 500 ppm. A more preferred concentration of a phenolic additive is a range from about 25 to about 200 ppm. A most preferred concentration of a phenolic additive is a range from about 50 to about 100 ppm.

20

          The concentration of phenolic additive of Formula (1) may be a single phenolic additive or a mixture of phenolic additives. The mixture of phenolic additive may be in any relative proportion.

25           Additives useful for the semiconductor industry are limited to species that are volatile, having a boiling point < 300 °C, and most preferably having a boiling point close to that of the product being stabilized. The boiling point of TMCTS is 135 °C, so it is most desirable to use an additive having a boiling point near or below 135 °C. Further, the application in the  
30   semiconductor industry will dictate the properties of the precursor selection. For example, formation of interlayer dielectric (ILD) films restricts precursor selection to use only silicon, oxygen, carbon and hydrogen due to compatibility issues with surrounding layers in a chip. The selection of

radical inhibitors must also follow this basis, so nitrogen, sulfur and phosphorous that are found in common radical inhibitors such as lecithin and lipoic acid must be avoided. Keeping in mind the volatility and elemental restrictions for inhibitors useful in the semiconductor industry, the preferred choices are phenol, hydroquinone, 4-methylphenol and 4-methoxyphenol, where the boiling points for these products range from 182 °C to 243 °C, respectively.

Examples of suitable alkoxysilane additive compounds are described by Formula (2) where  $R^6$  is a  $C_1$ - $C_{18}$  linear, cyclic or branched alkyl or substituted or unsubstituted aryl;  $R^7$ ,  $R^8$ , and  $R^9$  are independently H,  $C_1$ - $C_{18}$  linear, cyclic or branched alkyl,  $C_1$ - $C_{18}$  linear, cyclic or branched alkoxy or substituted or unsubstituted aryl. Examples of suitable  $R^6$  groups include, but are not limited to, methyl, ethyl, propyl, iso-propyl, butyl, iso-butyl, sec-butyl, tert-butyl, phenyl or methylphenyl. Examples of suitable  $R^7$ ,  $R^8$ , and  $R^9$  groups include, but are not limited to, H, methyl, ethyl, propyl, iso-propyl, butyl, iso-butyl, sec-butyl, tert-butyl, methoxy, ethoxy, propoxy, iso-propoxy, butoxy, iso-butoxy, sec-butoxy, tert-butoxy, phenyl or methylphenyl. Preferred examples of  $R^6$  in Formula (2) include, but are not limited to, methyl, ethyl and propyl. The most preferred example of  $R^6$  is methyl. Preferred examples of  $R^7$ ,  $R^8$  and  $R^9$  include, but are not limited to, methyl, ethyl, propyl, methoxy, ethoxy and propoxy. Most preferred examples of  $R^6$  are methyl and ethyl. Most preferred examples of  $R^7$ ,  $R^8$  and  $R^9$  include methyl, ethyl, methoxy and ethoxy.

Suitable examples of Formula (2) include but are not limited to trimethylmethoxysilane, triethylmethoxysilane, tripropylmethoxysilane, triphenylmethoxysilane, tri(4-methylphenyl)methoxysilane, dimethyldimethoxysilane, diethyldimethoxysilane, dipropyldimethoxysilane, diphenyldimethoxysilane, di(4-methylphenyl)dimethoxysilane, methyltrimethoxysilane, ethyltrimethoxysilane, propyltrimethoxysilane, phenyltrimethoxysilane, 4-methylphenyltrimethoxysilane, trimethylethoxysilane, triethylethoxysilane, tripropylethoxysilane,

triphenylethoxysilane, tri(4-methylphenyl)ethoxysilane,  
dimethyldiethoxysilane, diethyldiethoxysilane, dipropyldiethoxysilane,  
diphenyldiethoxysilane, di(4-methylphenyl)diethoxysilane,  
methyltriethoxysilane, ethyltriethoxysilane, propyltriethoxysilane,  
5 phenyltriethoxysilane, 4-methylphenyltriethoxysilane,  
trimethylpropoxysilane, triethylpropoxysilane, tripropylpropoxysilane,  
triphenylpropoxysilane, tri(4-methylphenyl)propoxysilane,  
dimethyldipropoxysilane, diethyldipropoxysilane, dipropyldipropoxysilane,  
diphenyldipropoxysilane, di(4-methylphenyl)dipropoxysilane,  
10 methyltripropoxysilane, ethyltripropoxysilane, propyltripropoxysilane,  
phenyltripropoxysilane, 4-methylphenyltripropoxysilane,  
trimethylbutoxysilane, triethylbutoxysilane, tripropylbutoxysilane,  
triphenylbutoxysilane, tri(4-methylphenyl)butoxysilane,  
dimethyldibutoxysilane, diethyldibutoxysilane, dipropyldibutoxysilane,  
15 diphenyldibutoxysilane, di(4-methylphenyl)dibutoxysilane,  
methyltributoxysilane, ethyltributoxysilane, propyltributoxysilane,  
phenyltributoxysilane, 4-methylphenyltributoxysilane,  
trimethylphenoxysilane, triethylphenoxysilane, tripropylphenoxysilane,  
triphenylphenoxysilane, tri(4-methylphenyl)phenoxysilane,  
20 dimethyldiphenoxysilane, diethyldiphenoxysilane, dipropyldiphenoxysilane,  
diphenyldiphenoxysilane, di(4-methylphenyl)diphenoxysilane,  
methyltriphenoxysilane, ethyltriphenoxysilane, propyltriphenoxysilane,  
phenyltriphenoxysilane, 4-methylphenyltriphenoxysilane, trimethyl(4-  
methylphenoxy)silane, triethyl(4-methylphenoxy)silane, tripropyl(4-  
25 methylphenoxy)silane, triphenyl(4-methylphenoxy)silane, tri(4-  
methylphenyl)(4-methylphenoxy)silane, dimethyldi(4-  
methylphenoxy)silane, diethyldi(4-methylphenoxy)silane, dipropyldi(4-  
methylphenoxy)silane, diphenyldi(4-methylphenoxy)silane, di(4-  
methylphenyl)di(4-methylphenoxy)silane, methyltri(4-methylphenoxy)silane,  
30 ethyltri(4-methylphenoxy)silane, propyltri(4-methylphenoxy)silane,  
phenyltri(4-methylphenoxy)silane and 4-methylphenyltri(4-  
methylphenoxy)silane. Preferred examples of Formula (2) include  
trimethylmethoxysilane, triethylmethoxysilane, tripropylmethoxysilane,

dimethyldimethoxysilane, diethyldimethoxysilane, dipropyldimethoxysilane, trimethylethoxysilane, triethylethoxysilane, tripropylethoxysilane, dimethyldiethoxysilane, diethyldiethoxysilane, dipropyldiethoxysilane, trimethylpropoxysilane, triethylpropoxysilane, tripropylpropoxysilane, 5 dimethyldipropoxysilane, diethyldipropoxysilane, dipropyldipropoxysilane, trimethylbutoxysilane, triethylbutoxysilane, tripropylbutoxysilane, dimethyldibutoxysilane, diethyldibutoxysilane and dipropyldibutoxysilane. Most preferred examples of Formula (2) include trimethylmethoxysilane, triethylmethoxysilane, tripropylmethoxysilane, trimethylethoxysilane, 10 triethylethoxysilane, tripropylethoxysilane, trimethylpropoxysilane, triethylpropoxysilane, tripropylpropoxysilane, trimethylbutoxysilane, triethylbutoxysilane, and tripropylbutoxysilane.

A suitable concentration of alkoxysilane additive compounds of 15 Formula (2) may be obtained from a range of concentrations, such as from about 1 ppm to about 5000 ppm. A suitable concentration range can be defined by defining a low end of suitable concentrations and a high end of suitable concentrations.

20 The low end of the suitable concentration ranges for the alkoxysilane additive compound may be from about 1 ppm to about 500 ppm. Suitable concentrations for the low end concentration range include but are not limited to 1 ppm, 5 ppm, 10 ppm, 25 ppm, 50 ppm, 100 ppm, 200 ppm, 300 ppm, 400 ppm and 500 ppm.

25 The high end of the suitable concentration ranges for the alkoxysilane additive compound may be from about 1000 ppm to about 5000 ppm. Examples of suitable concentrations for the high-end concentration range include but are not limited to 1000 ppm, 1500 ppm, 30 2000 ppm, 2500 ppm, 3000 ppm, 3500 ppm, 4000 ppm, 4500 ppm, and 5000 ppm.

The concentration of alkoxysilane additive of Formula (2) may be a single alkoxysilane additive or a mixture of alkoxysilane additives, and the mixture of alkoxysilane additives may be in any relative proportion.

5           A preferred concentration when both alkoxysilane additives and phenolic additives are employed, varies based on the identity of the siloxane, the phenolic additive and the alkoxysilane additive. In general, it is preferred to have the concentration of the alkoxysilane exceed that of the phenolic additive when the two are used in combination. This is due to  
10   solubility issues that can be encountered when using a greater percentage of solid phenolic additive, especially in higher molecular weight organohydrosiloxanes. Further, it can be beneficial to dissolve a solid, phenolic additive in an alkoxysilane prior to stabilizing an organohydrosiloxane. This allows for addition of a single, homogeneous  
15   and easily metered liquid without the concern for having problems with washing solids from container to another.

          An example of a suitable concentration of alkoxysilane additive is a range from about 1 to about 5000 ppm in addition to a suitable  
20   concentration of phenolic additive being from about 1 to about 1000 ppm. An example of a preferred concentration of alkoxysilane additive is a range from about 10 to about 2500 ppm in addition to a preferred concentration of phenolic additive being from about 10 to about 500 ppm. An example of a more preferred concentration of alkoxysilane additive is a range from about  
25   100 to about 1000 ppm in addition to a more preferred concentration of phenolic additive being from about 25 to about 200 ppm. An example of a most preferred concentration of alkoxysilane additive is a range from about 250 to about 500 ppm in addition to a most preferred concentration of phenolic additive being from about 50 to about 100 ppm.

30

          Referring to Figure 1, chemical precursors of the type described by Formulas (3), (4) and (5) are typically packaged, shipped and stored in stainless steel containers (1) to retain product quality for the maximum

amount of time. To extend shelf life of a precursor, it is common to package the product under an inert gas such as nitrogen, argon or helium. The product container is then connected to chemical delivery equipment that transfers the chemical by a precisely controlled means, to retain  
5 product and process purity and consistency, to process equipment, referred to here as a film deposition tool.

The precursor is transported from the container (1), through chemical delivery lines (2), to a vaporizer or misting apparatus (3) housed  
10 in the film deposition tool (11). The precursor can be transported from the container (1) through the delivery line (2) to the vaporizer or misting apparatus (3) by a variety of means, including, but not limited to, pressurization of the container with an inert gas, use of a mechanical pumping mechanism, gravity feed, or combinations thereof.

15 A suitable precursor flow rate can range from about 0.01 to about 10 mL/minute. The vaporizer or misting apparatus (3) serves as a means to rapidly convert liquid precursor to a vapor or mist, and it can use various techniques, such as heat, high pressure gas, or other means to accomplish  
20 this task, though conversion of the precursor from a liquid to a gaseous state may take place either in the vaporizer or misting apparatus (3) or in the chemical vapor process line (5). The precursor is injected in the form of a vapor or mist into the chemical vapor process line (5) that is commonly heated between about 30 °C and about 120 °C to prevent the precursor  
25 vapor from condensing inside the line (5). This line is connected to the process chamber (4) inside the film deposition tool (11), and substrate (6) is housed within the process chamber (4). The process chamber (4) and chemical vapor process line (5) may be operated at ambient pressure (i.e., 760 torr), but it is also common to operate below atmospheric pressure,  
30 from about 0.01 torr to about 700 torr, to enhance vaporization of the precursor and to help keep the precursor in the vapor phase.



The precursor vapor is transported through the chemical vapor process line (5) to the substrate (6) in the process chamber (4) by a stream of gas flowing past the vaporizer or misting apparatus (3). The stream of gas is supplied from a source tank (7) and flows through a gas delivery line (8) to the chemical vapor process line (5). The stream of gas, having a flow rate of about 5 sccm to about 10,000 sccm, is often heated to enhance vaporization of the precursor and to help keep the precursor in the vapor phase. The gas used may be inert, such as nitrogen, helium or argon, chosen simply to act as a means to transport the precursor vapor to the substrate, or it may be a reactive gas, such as oxygen, ozone, ammonia, nitrous oxide, carbon dioxide, carbon monoxide,  $\text{SiH}_4$ , silanes, silicon tetrafluoride, hydrazine and the like. Certain reactive gases containing elements not commonly used in the deposition of a low k film, such as ammonia that contains nitrogen, may be used in situations where modification of a deposited film is desired. It is not uncommon to complete the deposition process of a low k film by doping the top of the film with nitrogen by bleeding ammonia gas into the final phase of the deposition process.

As the precursor vapor is transported to the substrate (6), it may be mixed with one or more reactants, in addition to the transport gas, to enhance its deposition onto the substrate or to dope the resulting film in a particular manner. The reactants may be reactive gases as mentioned above, or they may be other chemical precursors such as amines, aminoalcohols, silanes, siloxanes, alkanes, alkenes, alkynes, alcohols, esters, ketones, aldehydes, carboxylic acids, arsines, alkylarsenates, phosphines, alkylphosphates, boranes, alkylborates, and the like, or any combinations thereof. By way of example, the following precursors may be used in the present invention: methylamine, dimethylamine, 2-(dimethylamino)ethanol, diethylsilane, vinylmethyldimethoxysilane, octane, cyclobutene, trimethylsilylacetylene, t-butanol, methylbenzoate, acetone, benzaldehyde or acetic acid, and dopant chemicals, such as, arsine, trimethylarsenate, triethylarsenate, phosphine, trimethylphosphate,

triethylphosphate, trimethylphosphite, triethylphosphite, diborane, trimethylborate, and triethylborate. The reactants are carefully selected to enhance the deposition of precursor on the substrate, and to modify the chemical identity and properties of the layer deposited onto the substrate.

- 5 These reactants can be introduced into the film deposition tool (11) by various means and at various locations in the process, depending on the desired effect. It is most convenient to introduce a reactant into the film deposition tool (11) in gaseous form, so it would be necessary to have a  
10 additional vaporizer or misting apparatus in the case where liquid reactants are used. An additional vaporizer or misting apparatus, or gas delivery line used to introduce a reactant can be placed near the point where the gas delivery line (8) meets the chemical vapor process line (5), upstream or downstream of the vaporizer or misting apparatus, directly into or near the plasma (9), and/or somewhere on the sides, top, or bottom of the film  
15 process chamber (4) of the film deposition tool (11).

- The precursor vapor, and potential reactants, may also experience other conditions used to enhance deposition, such as heat or plasma (9). The precursor vapor may be preheated to between about 100 °C and about  
20 800 °C before contact with the substrate to enhance the deposition of the precursor on the substrate. A plasma may also be used to add energy to the precursor vapor and enhance the deposition. Additionally, the plasma may be pulsed on and off to change the properties of the deposited film. The plasma power and pulse duration are carefully selected to enhance the  
25 deposition of precursor on the substrate, and to modify the chemical identity and properties of the layer deposited onto the substrate. The plasma may also be applied over a range of frequencies, where the high and low frequency plasma power may range from about 0 to several kilowatts. The substrate may also have a bias of between about 0 and  
30 about –400 VDC to enhance material transport to the substrate. The substrate may be heated from about 25 °C to about 500 °C to either cause thermal breakdown of the precursor on the substrate, or may be used to enhance the deposition of precursor on the substrate.

If only organohydrosiloxane and inert gas are used in this deposition process, the resulting film will be composed of Si, O, C and H. If oxygen is used in the gas stream, the amount of oxygen in the film can be increased.

5 Carbon and nitrogen can be added by using carbon or nitrogen containing reactants. In addition, arsenic, boron and phosphorous can be added by using reactants containing these compounds, as previously described. The film that is deposited onto a substrate is wholly dependent on the reactants that contact it. Unreacted materials are exhausted through an exhaust line

10 (10).

### EXAMPLES

A series of tests were devised to compare the decomposition of pure

15 TMCTS under various conditions with TMCTS that has been stabilized with additives. In these tests, approximately 5 grams of stabilized or unstabilized TMCTS sample was placed in a glass ampoule. To remove adventitious gas impurities from the sample before it was sealed, the sample was frozen in liquid nitrogen, the headspace of the ampoule was

20 evacuated under dynamic vacuum, and then the sample was thawed under static vacuum. This process was repeated, and the sample was then flame sealed with or without a reactive gas. Control samples were sealed under vacuum, while reactive test samples were sealed under approximately 0.5 wt% of a reactive gas at ambient pressure.

25

The reactive gases that were used were oxygen and carbon dioxide. These gases were chosen because they are reactive components of air. Humid air was also used, and it was generated by flowing normal room air through a water bubbler in order to saturate the air with water vapor. This

30 condition was used to simulate the worst-case scenario of exposing TMCTS to air at 100% humidity.

Samples were prepared for each potential reactive condition, and they were then exposed to a range of temperatures from 25 to 120 °C for a period of five days. After the experiment was complete, total decomposition was measured in each sample by gas chromatography (GC). The percentage of decomposition was derived by comparing gas chromatograms of samples taken before and after each experiment, where an increase in the concentration of any other species in the chromatogram indicated decomposition.

## 10 **EXAMPLES 1-20**

Results outlined in Table 1 show that for pure TMCTS, a small amount of decomposition was observed with samples heated above 60 °C when exposed to carbon dioxide or kept under vacuum. When samples were exposed to oxygen or air, more significant decomposition was seen at all temperatures. In similar experiments using 5,000 ppm trimethylmethoxysilane (TMMOS) as a stabilizing agent, no decomposition was observed from 60 to 120 °C under vacuum or with exposure to carbon dioxide. This indicates that the TMMOS stabilizer prevented decomposition when compared to pure TMCTS under similar conditions. However, TMMOS did not stabilize TMCTS in the presence of oxygen.

Table 1

Example #	T (C)	Reactive Gas	Pure TMCTS	TMCTS + 0.5 wt% TMMOS	TMCTS + 100 ppm MHQ	TMCTS + 100ppm MHQ + 0.5 wt% TMMOS
			% Decomp	% Decomp	% Decomp	% Decomp
1	25	Vacuum	0	0	0	0
2	60	Vacuum	0.04	0	0	0
3	80	Vacuum	0.01	0	0	0
4	90	Vacuum	0.02	0	0	0
5	120	Vacuum	0.03	0	0.06	0.01
6	25	CO <sub>2</sub>	0	0	0	0
7	60	CO <sub>2</sub>	0.02	0	0	0
8	80	CO <sub>2</sub>	0.02	0	0	0
9	90	CO <sub>2</sub>	0.02	0	0	0
10	120	CO <sub>2</sub>	0.01	0.01	0.01	0.02
11	25	O <sub>2</sub>	0.04	0.00	0	0
12	60	O <sub>2</sub>	4.94	5.64	0	0
13	80	O <sub>2</sub>	4.83	5.17	0	0
14	90	O <sub>2</sub>	4.87	6.21	0	0
15	120	O <sub>2</sub>	5.28	6.39	0.16	0.05
16	25	Air	1.00	No Data	No Data	0
17	60	Air	1.19	No Data	No Data	0
18	80	Air	1.25	No Data	No Data	0
19	90	Air	1.29	No Data	No Data	0
20	120	Air	0.98	No Data	No Data	0.06

The results in Table 1 show that, when 4-methoxyphenol (MHQ) was used, it proved to be a very effective stabilizer at all temperatures, with no observable decomposition of TMCTS up to 90 °C, and only minimal decomposition at 120 °C. The results also demonstrate that the use of MHQ in combination with TMMOS gave slightly better results at 120 °C. This is most apparent in Example 15, where the use of MHQ and TMMOS reduced the total percent decomposition from 0.16% to 0.05%.

#### EXAMPLES 21-27

A range of additive concentrations was then tested to determine the minimum amount of additive necessary to stabilize TMCTS effectively. Samples of TMCTS were prepared with concentrations of MHQ from 0 to 5000 ppm, and the samples were sealed in glass ampoules under an

oxygen ambient as described above. The samples were heated at 120 °C for five days, and then the total decomposition was measured in each sample by GC. The percentage of decomposition was derived by comparing gas chromatograms of samples taken before and after each experiment, where an increase in the concentration of any other species in the chromatogram indicated decomposition. The results shown in Table 2 demonstrate that a concentration of MHQ as low as 10 ppm effectively stabilizes TMCTS against decomposition. Concentrations of MHQ up to 5000 ppm were shown to be effective in stabilizing TMCTS, although solubility issues were observed at concentrations of 1000 ppm or greater of MHQ. While these concentrations are soluble in TMCTS, it becomes increasingly difficult to initially dissolve MHQ in TMCTS as its concentration increases.

**Table 2**

Example #	T (C)	Reactive Gas	Quantity of MHQ	% Decomp
21	120	O <sub>2</sub>	0 ppm	5.28
22	120	O <sub>2</sub>	10 ppm	0.66
23	120	O <sub>2</sub>	100 ppm	0.16
24	120	O <sub>2</sub>	500 ppm	0.05
25	120	O <sub>2</sub>	1000 ppm	0.06
26	120	O <sub>2</sub>	2500 ppm	0.12
27	120	O <sub>2</sub>	5000 ppm	0.20

**EXAMPLES 28-36**

Two more additive types were tested to determine their effectiveness to stabilize TMCTS, and these differed by varying substituents and substituent positions on a phenolic ring. Samples of TMCTS were prepared for each reactive condition using 100 ppm of the two different phenolic additives, phenol and cresol. The samples were heated at 120 °C for five days. The total decomposition was measured in each sample by GC. The percentage of decomposition was derived by comparing gas chromatograms of samples taken before and after each experiment, where an increase in the concentration of any other species in the chromatogram indicated decomposition. These results are compared

beside results from Table 1 for pure TMCTS and TMCTS with 100 ppm of MHQ to demonstrate that phenol and m-cresol stabilize TMCTS equally as well as MHQ in the presence of oxygen and moist air.

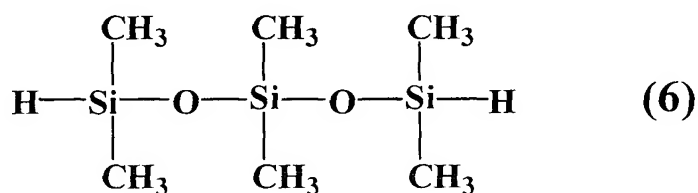
Table 3			Pure TMCTS	100 ppm MHQ	100 ppm Phenol	100 ppm Cresol
Example #	T (C)	Reactive Gas	% Decomp	% Decomp	% Decomp	% Decomp
28	25	Vacuum	0	0	0	0
29	60	Vacuum	0.04	0	0.12	0
30	90	Vacuum	0.02	0	0.01	0.03
31	25	O <sub>2</sub>	0.04	0	0	0
32	60	O <sub>2</sub>	4.94	0	0	0
33	90	O <sub>2</sub>	4.87	0	0	0.03
34	25	Humid Air	1.00	No Data	0	0
35	60	Humid Air	1.19	No Data	0	0
36	90	Humid Air	1.29	No Data	0.03	0.05

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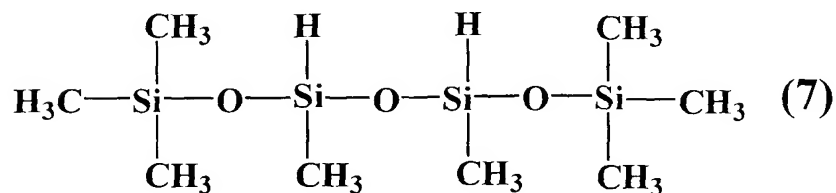
#### EXAMPLES 37-45

To demonstrate the ability to stabilize linear siloxanes, two types of linear siloxanes were chosen for further testing. 1,1,3,3,5,5-hexamethyltrisiloxane (HMTS), shown in Formula (6), was chosen for its terminal hydrogens on the terminal silicon atoms. 1,1,1,3,5,7,7,7-octamethyltetrasiloxane (OMTS), shown in Formula (7), was chosen for its internal hydrogens on central silicon atoms.

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Samples of HMTS and OMTS were prepared for each reactive condition either without additives, or using 100 ppm of MHQ as shown in Table 4. The samples were heated at 120 °C for five days. The total decomposition was measured in each sample by GC. The percentage of decomposition was derived by comparing gas chromatograms of samples taken before and after each experiment, where an increase in the concentration of any other species in the chromatogram indicated decomposition. These results are compared to results from Table 1 for pure TMCTS and TMCTS with 100 ppm of MHQ. The data demonstrates that decomposition also occurs when linear organohydrosiloxanes are exposed to oxygen or humid air and that each type of linear organohydrosiloxane was stabilized using 100 ppm of MHQ.

Table 4			Pure TMCTS	TMCTS + 100 ppm	Pure HMTS	HMTS + 100 ppm	Pure OMTS	OMTS + 100 ppm
Exam. #	T (C)	Reactive Gas	% Decomp	% Decomp	% Decomp	% Decomp	% Decomp	% Decomp
37	25	Vacuum	0	0	0	0	0.56	0
38	60	Vacuum	0.04	0	0.02	0.02	0.53	0
39	90	Vacuum	0.02	0	0.02	0	0.38	0
40	25	O <sub>2</sub>	0.04	0	0	0	0	0
41	60	O <sub>2</sub>	4.94	0	0	0	0	0
42	90	O <sub>2</sub>	4.87	0	2.65	0.01	5.51	0
43	25	Humid Air	1.00	No Data	0.14	0	0.29	0
44	60	Humid Air	1.19	No Data	0.10	0	0	0
45	90	Humid Air	1.29	No Data	0.11	0	0.58	0

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#### EXAMPLE 46

The precursor, 1,3,5,7-tetramethylcyclotetrasiloxane, stabilized with 100 ppm of 4-methoxyphenol and 150 ppm of trimethylmethoxysilane is transferred, with the aid of nitrogen pressure, from a stainless steel container through a chemical delivery line to a fuel-injector style vaporizer at a flow rate of 5 mL/min. The precursor is vaporized into a chemical vapor process line that is heated to 80 °C and transported to a substrate using a reactive gas mixture composed of 1,000 sccm of carbon dioxide and 125 sccm of oxygen with the system base pressure held at 60 torr.



During transport to the substrate, the precursor vapor and reactive gases are exposed to a high frequency plasma power of 1,500 W and a low frequency plasma power of 250 W. The substrate is heated to 250 °C with a substrate bias of -15 VDC. A carbon doped silicon oxide film is  
5 deposited on the substrate using these conditions.

While we have shown and described several embodiments in accordance with our invention, it is to be clearly understood that the same are susceptible to numerous changes apparent to one skilled in the art.  
10 Therefore, we do not wish to be limited to the details shown and described but intend to show all changes and modifications that come within the scope of the appended claims.